

جامعة الزيتونة الأردنية Al-Zaytoonah University of Jordan كلية العلوم وتكنولوجيا المعلومات Faculty of Science and information Technology

" عراقة وجودة" "Tradition and Quality"

فكر حضاري وحوار متمدن Civilized Thought ...Civilized

Dialogue

QF04/0408-4.0E

Course Plan for Bachelor program - Study Plan Development and Updating Procedures/ Department of Basic Sciences

Study plan No.	2024/2025		University Specialization		Bachelor of Physical Therapy	
Course No.	0420821		Course name		Practical General Chemistry	
Credit Hours	1		Prerequisite/ Co-requisite		*General Chemistry For Medical Sciences	
Course type	□ MANDATORY UNIVERSITY REQUIREMENT	□ UNIVERSITY ELECTIVE REQUIREMENTS	☐ FACULTY MANDATORY REQUIREMENT	□ Support course family requirements	□ Mandatory requirements	□ Elective requirements
Teaching style	□ Full online learning		□ Blended learning		□ Traditional learning	
Teaching model	□ 1 Synchronous: 1 asynchronous		□ 1 face to face : 1 asynchronous		□ 1 Traditional	

Faculty member and study divisions' information (to be filled in each semester by the subject instructor)

Name	Academic rank	Office No.	Phone No.	E-n	nail
Division number	Time	Place	Number of students	Teaching style	Approved model
				Traditional Learning	1 Traditional

Brief description

This course covers the practical applications of the most important theoretical concepts covered in the General Chemistry course, such as qualitative and quantitative studies, stoichiometry, volumetric analysis, and thermochemical changes.

Learning resources

Course book information	Laboratory Manual for General Chemistry, Prepared by M. Sc. Sawsan			
(Title, author, date of issue,	Shraim 2021 Al-Zaytoonah University of Jordan			
publisher etc)	Sinann, 2021, In Zaytoonan Oniversity of Jordan			
Supportive learning resources	1. Chemistry in the Laboratory, James M. Postma (California State			
(Books, databases,	University, Chico), Julian L. Roberts (University of Redlands), Anne			
periodicals, software,	Roberts, 8 th edition, 2017			
applications, others)	2. Chemistry, Th	e Central Science	e, Brown, Le May	, Bursten Prentice
	Hall, 14th editi	on (2017).	•	
	3. Chemistry, by Raymond Chang, Kenneth Goldsby, 12 th edition, AP			
	student edition, 2016.			
Supporting websites				
The physical environment for	□ Class	□ labs	□ Virtual	□ Others
teaching	room		educational	
			platform	
Necessary equipment and	Moodle			
software				
Supporting people with	-			



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 special needs
 E-Learning & Open Educational Resources Center. Email: elearning@zuj.edu.jo; Phone: +962 6 429 1511 ext. 425/362.

Course learning outcomes (S = Skills, C = Competences K = Knowledge,)

No.	Course learning outcomes	The associated program learning output code			
	Knowledge				
The s	tudent should be able to:				
K1	Recognize the proper basics of safe lab work.	MK1			
K2	Outline standard laboratory procedures.	MK1			
K3	Report observations and results.	MK1			
	Skills				
	The student should be able to:				
S1	Perform lab procedures for experiments covered in this course and	MS4			
	present the results.				
S2	Interpret data and observations obtained from performed experiments.	MS4			
S3	Use instruments, glassware and chemicals properly and safely.	MS4			
S4	Manage the risks of chemical substances and procedures.	MS4			
	Competences				
C1	Develop his/her professional and personal performance by				
	continuously attending labs, submitting reports on time, and work	MC3			
	effectively within groups.				

Mechanisms for direct evaluation of learning outcomes

Type of assessment / learning style	Fully electronic learning	Blended learning	Traditional Learning (Theory Learning)	Traditional Learning (Practical Learning)
Midterm exam	30%	30%	30%	0%
Participation / practical applications	0	0	0	60%
Asynchronous interactive activities	30%	30%	30%	0%
Final exam	40%	40%	40%	40%

Note 1: Asynchronous interactive activities are activities, tasks, projects, assignments, research, studies, projects, work within student groups ... etc, which the student carries out on his own, through the virtual platform without a direct encounter with the subject teacher.

Note 2: According to the Regulations of granting Master's degree at Al-Zaytoonah University of Jordan, 40% of final evaluation goes for the final exam, and 60% for the semester work (examinations, reports, research or any scientific activity assigned to the student).



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Schedule of simultaneous / face-to-face encounters and their topics

Week	Subject	learning style*	Reference **
1	Introduction and Check-in		
2	Instructions and Lab. Equipment	Lecture, lab-based learning	Lab Manual Page 1
3	Measurements and Chemical Observation	Lecture, lab-based learning	Lab Manual Page 8
4	Stoichiometry-I	Lecture, lab-based learning	Lab Manual Page 12
5	Stoichiometry-II	Lecture, lab-based learning	Lab Manual Page 17
6	Limiting Reactant	Lecture, lab-based learning	Lab Manual Page 24
7	Determination of an Unknown		
8	Volumetric Analysis (I): Acid-Base Titrations	Lecture, lab-based learning	Lab Manual Page 31
9	Volumetric Analysis (II): Redox Titrations	Lecture, lab-based learning	Lab Manual Page 39
10	Chemical Equilibrium	Lecture, lab-based learning	Lab Manual Page 45
11	Thermochemistry [Determination of	Lecture, lab-based learning	Lab Manual Page 50
12	Spectrophotometric Determination of the Solubility of NiSO4.6H2O	Lecture, lab-based learning	Lab Manual Page 57
13	Determination of an Unknown		
14	Check-out		
15	-		
16	Final Exam		

* Learning styles: Lecture, flipped learning, learning through projects, learning through problem solving, participatory learning ... etc.

** Reference: Pages in a book, database, recorded lecture, content on the e-learning platform, video, website ... etc.

Schedule of asynchronous interactive activities (in the case of e-learning and blended learning)

Week	Task / activity	Reference	Expected results
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